

Cadmium chloride

Product Stewardship Summary

Chemical Name:	Cadmium chloride
Synonyms:	Dichlorocadmium, Caddy
CAS Number:	10108-64-2
EC (EINECS) Number:	233-296-7
Revision number:	1-2024

- Chemical identification and uses: Cadmium Chloride is a whitish, crystalline, inorganic compound that forms toxic fumes of cadmium oxides when heated. Cadmium chloride is used in electroplating, printing, photocopying, dyeing, mirrors, vacuum tubes, lubricants, analytical chemistry and as a chemical intermediate to produce cadmium containing pigments and stabilizers.
- **Potential exposures:** Exposure to Cadmium chloride can occur in industrial/manufacturing facilities and/or during use as laboratory chemicals in research settings. The substance can be absorbed into the body by inhalation of its aerosol, through the skin and by ingestion. Good manufacturing and industrial hygiene practices for Cadmium sulfate should be followed to prevent or reduce contact. See the Safety Data Sheet (SDS) for additional information.
- Human Health hazards: Cadmium chlrode is toxic if swallowed or if inhaled. It may damage fertility or the unborn child, cause cancer or genetic defects. It causes damage to organs through prolonged or repeated exposure. Cadmium compounds are identified as known or suspected carcinogen by NTP, IARC, or ACGIH. These chemicals are known to the State of California to cause cancer and birth defects or other reproductive harm. One should refer to See the Safety Data Sheet (SDS) for additional information and any protective information.
- Environmental Health hazards: It is very toxic to aquatic life, may cause long-lasting adverse effects on the aquatic life.
- Please contact us at <u>ESSPSCustomerCare@Honeywell.com</u> for more information. Additional information Cadmium sulfate may also be found at the following links:

<u>Toxicological profile of Cadmium</u> Cadmium Chloride- Pubchem